

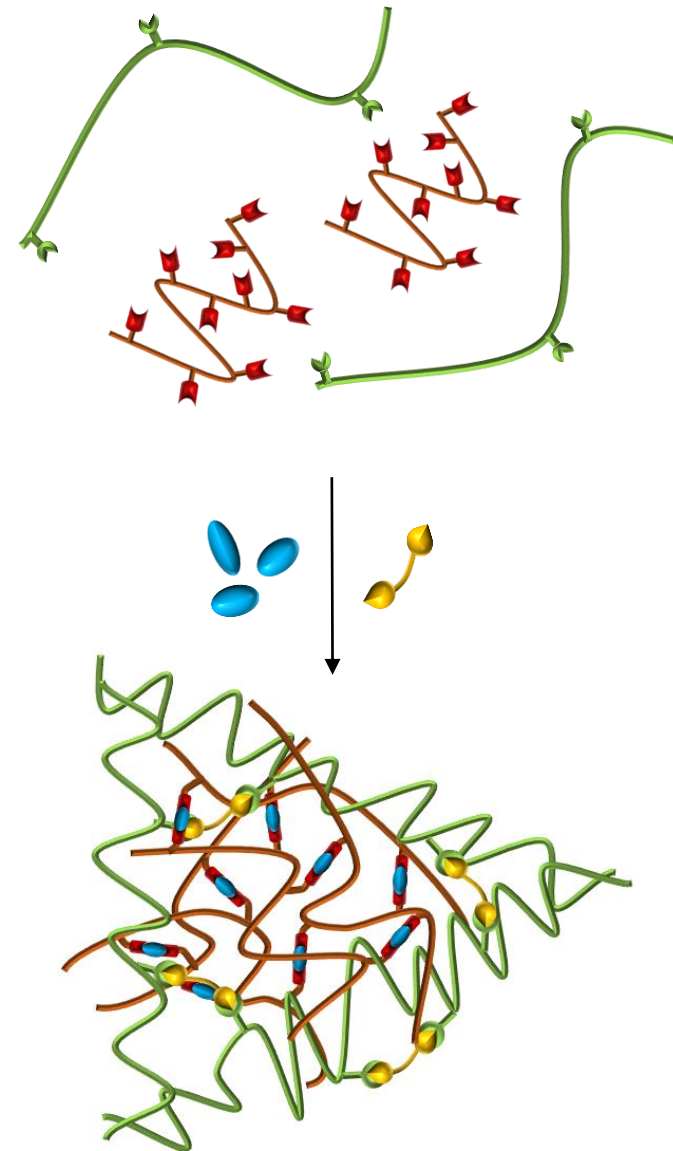


ICODnet
INITIAL TRAINING NETWORK

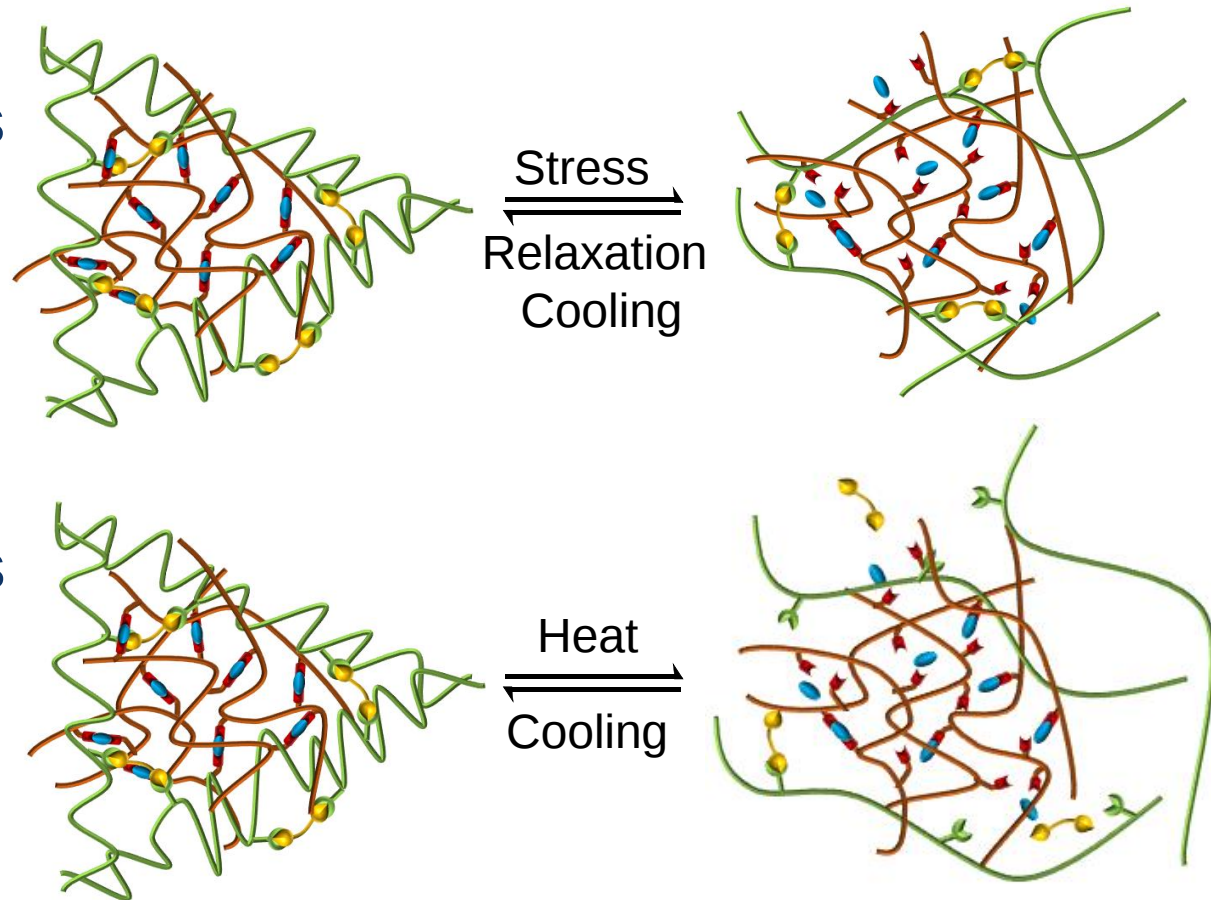
UPDATE DSM

Larissa Hammer, ESPCI Paris

- Design and synthesize DDNs based on reversible covalent bonds
- Creating two interpenetrating network structures
- Creating thermoplastics with functional groups
- Orthogonal cross-linking

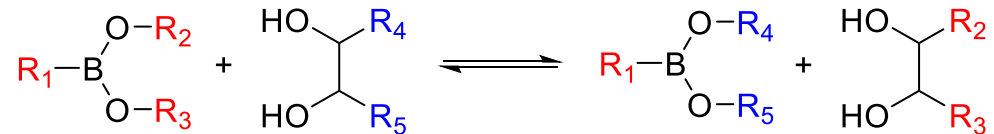


- Network 1: Brittle, sacrificial bonds, very dynamic
- Network 2: Soft, dynamic at high T
- Network 1 dissipates energy
- Heals itself
- Network 2 only gives up form at high T
- Recyclability

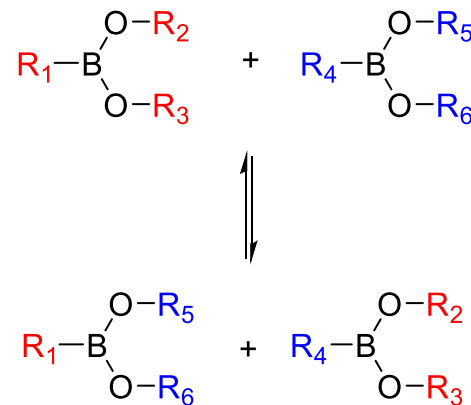


- Dynamic, brittle network
 - Sacrificial
 - Self-healing through dynamic covalent bonds
-
- Boronic ester exchange
 - Works at rt/ in bulk
 - 'Vitrimers Network'
 - Maintains same number of crosslinks
 - Self-healing, processable

Boronic ester / diol exchange:

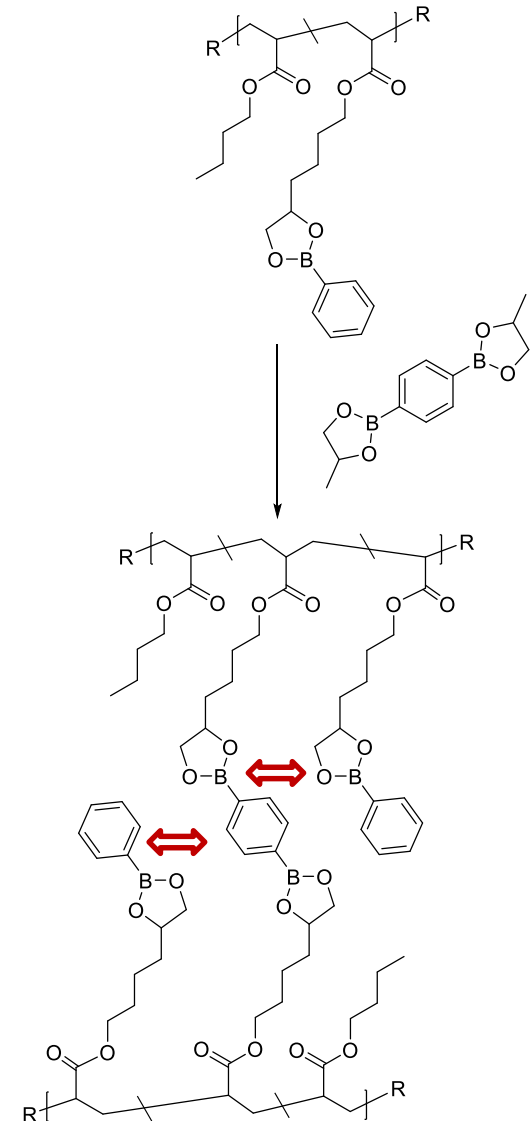


Boronic ester exchange:



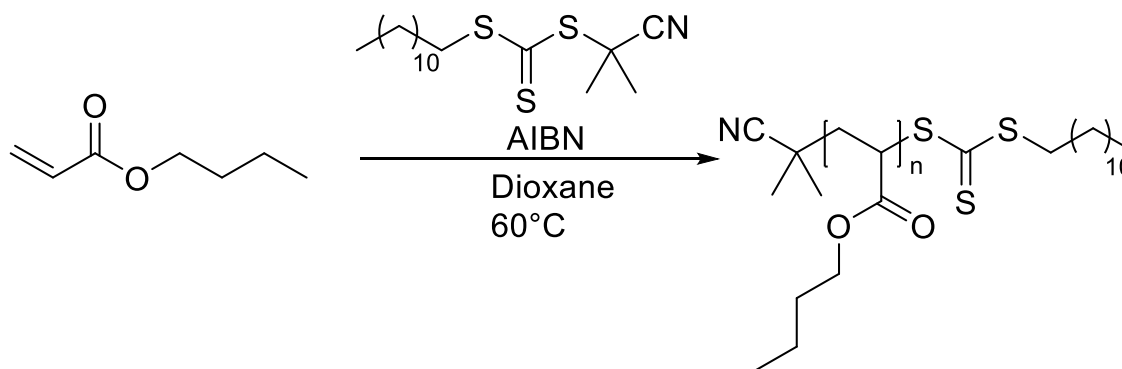
Exchange
accelerates in
presence of
diols

- Butyl acrylate system (low T_g)
- High density of cross links
 - DP 1000 (~100 000 Da)
 - ~100 links/chain (10% functionality)
- Diol-based polymer protected with boronic acid
- Cross-linking via di-boronic ester



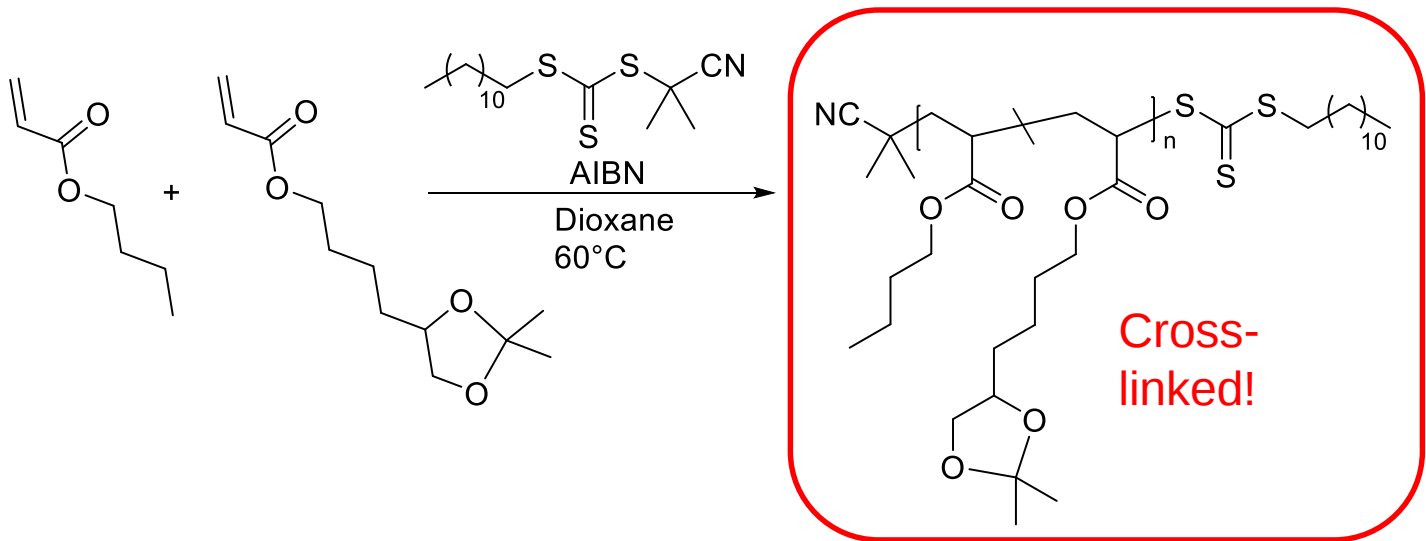
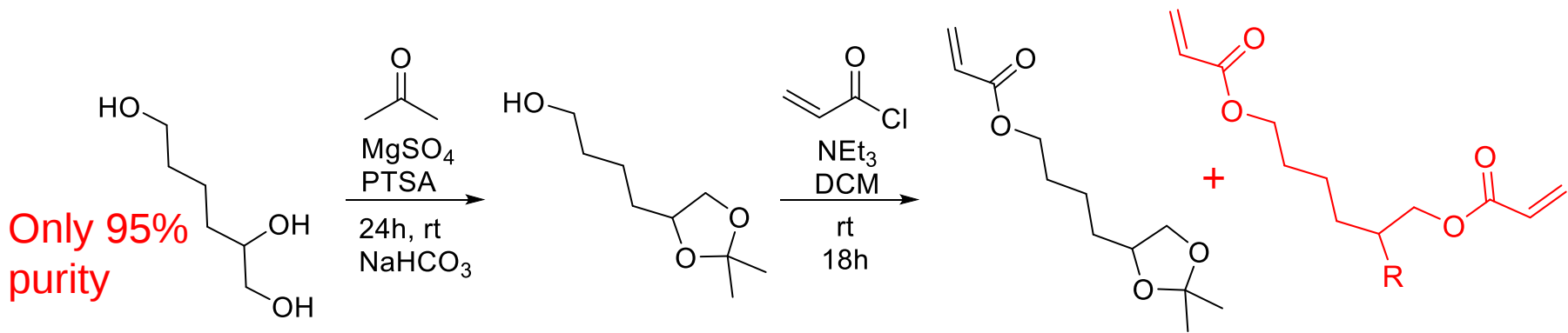
Test Polymerization

- Free radical polymerization $\rightarrow \bar{D} \sim 3$
- Reversible addition–fragmentation chain-transfer (RAFT) polymerization
- Looking for right conditions (Agent, T, solvent...)
- RAFT agent: CPDT¹⁾



- $M_n = 84,000$ Da, $\bar{D} = 1.20$

- Monomer synthesis for diol monomer

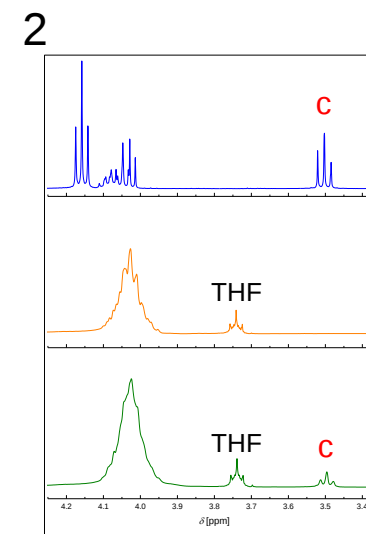
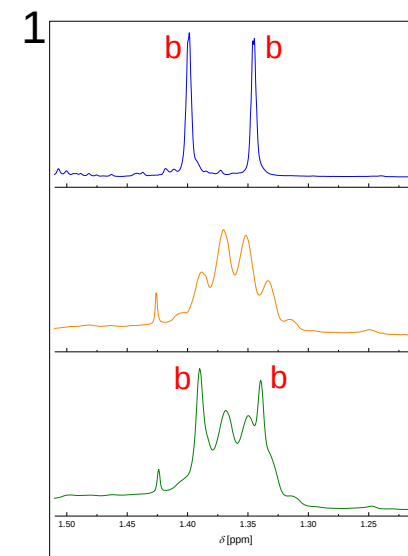
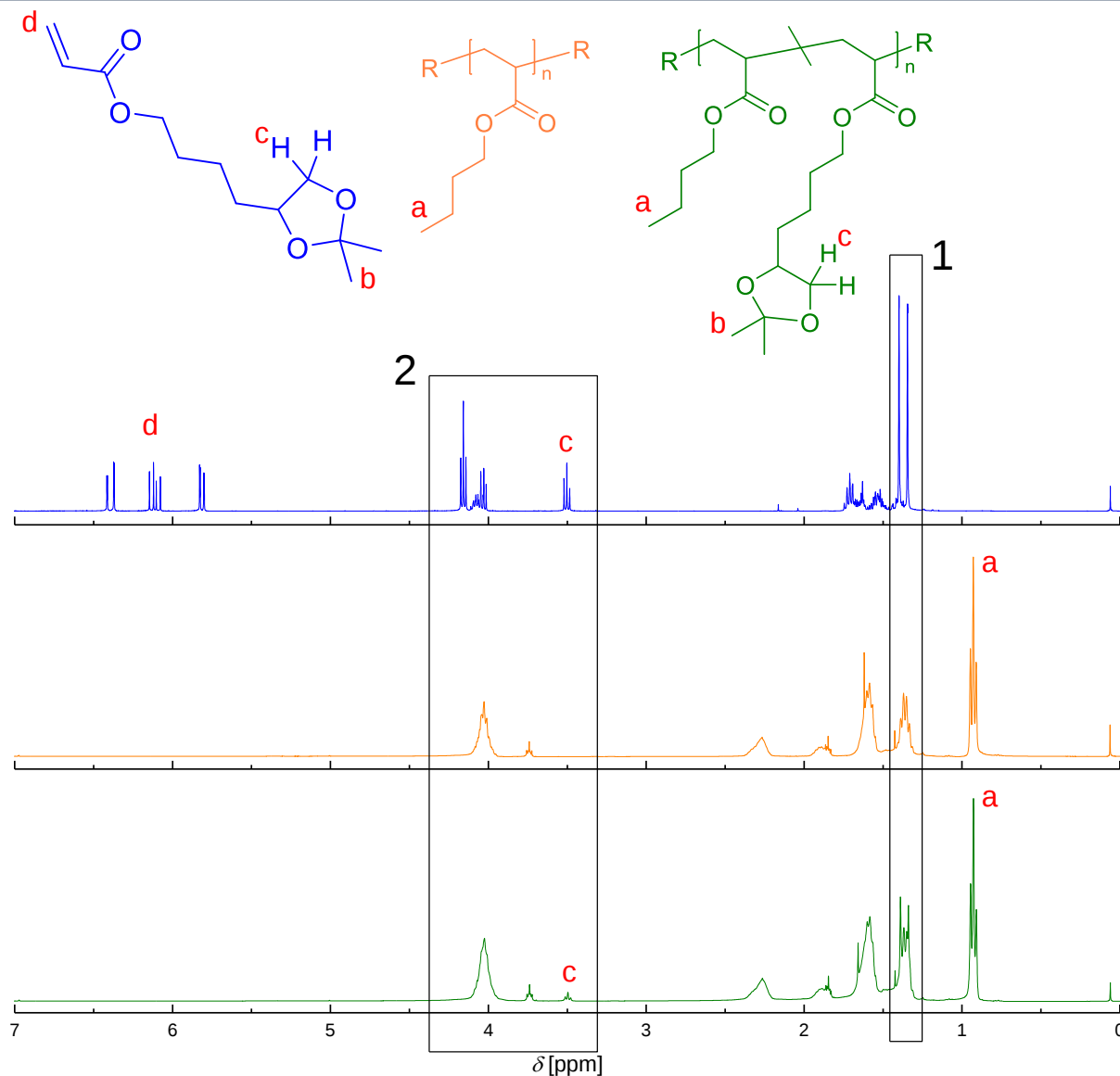


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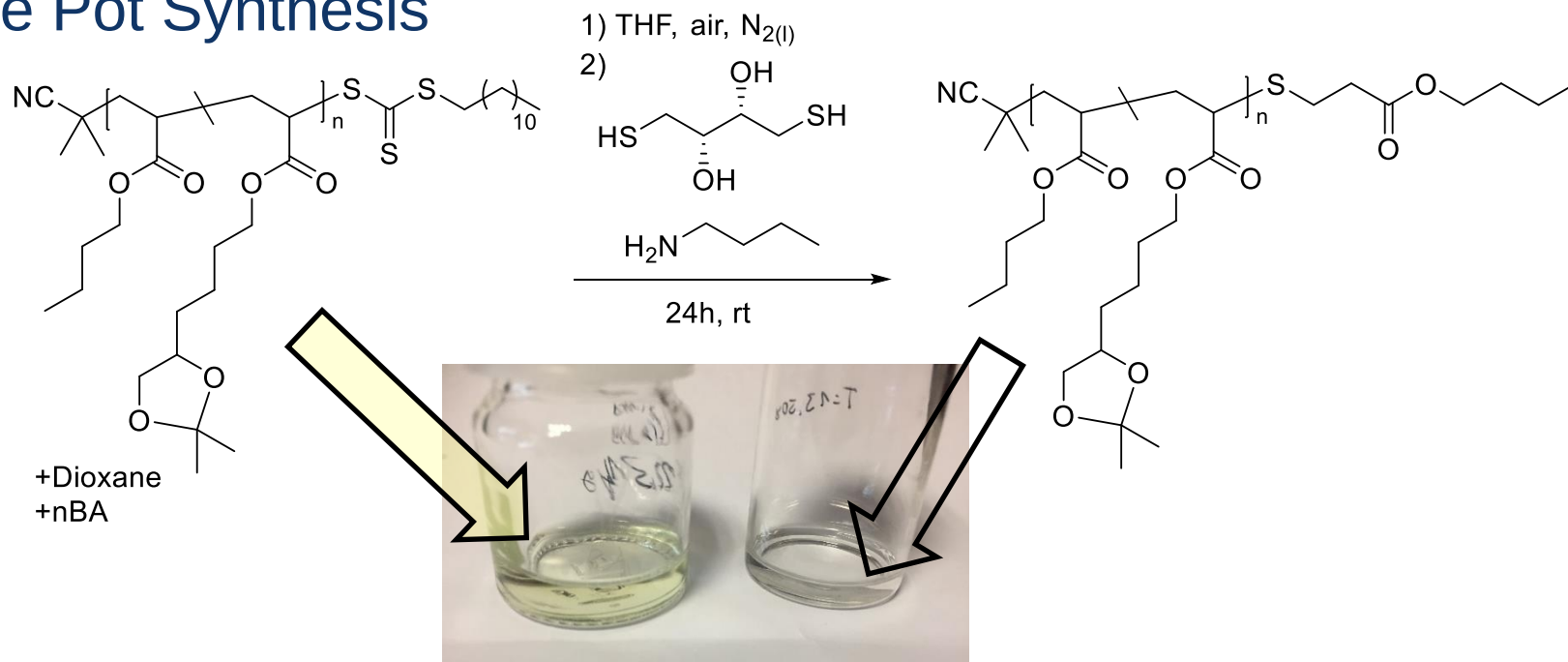


- ESPCI  PARIS | PSL 

Diol Copolymer NMR

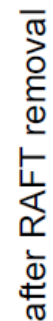


- RAFT group removal: aminolysis/Michael addition
- One Pot Synthesis

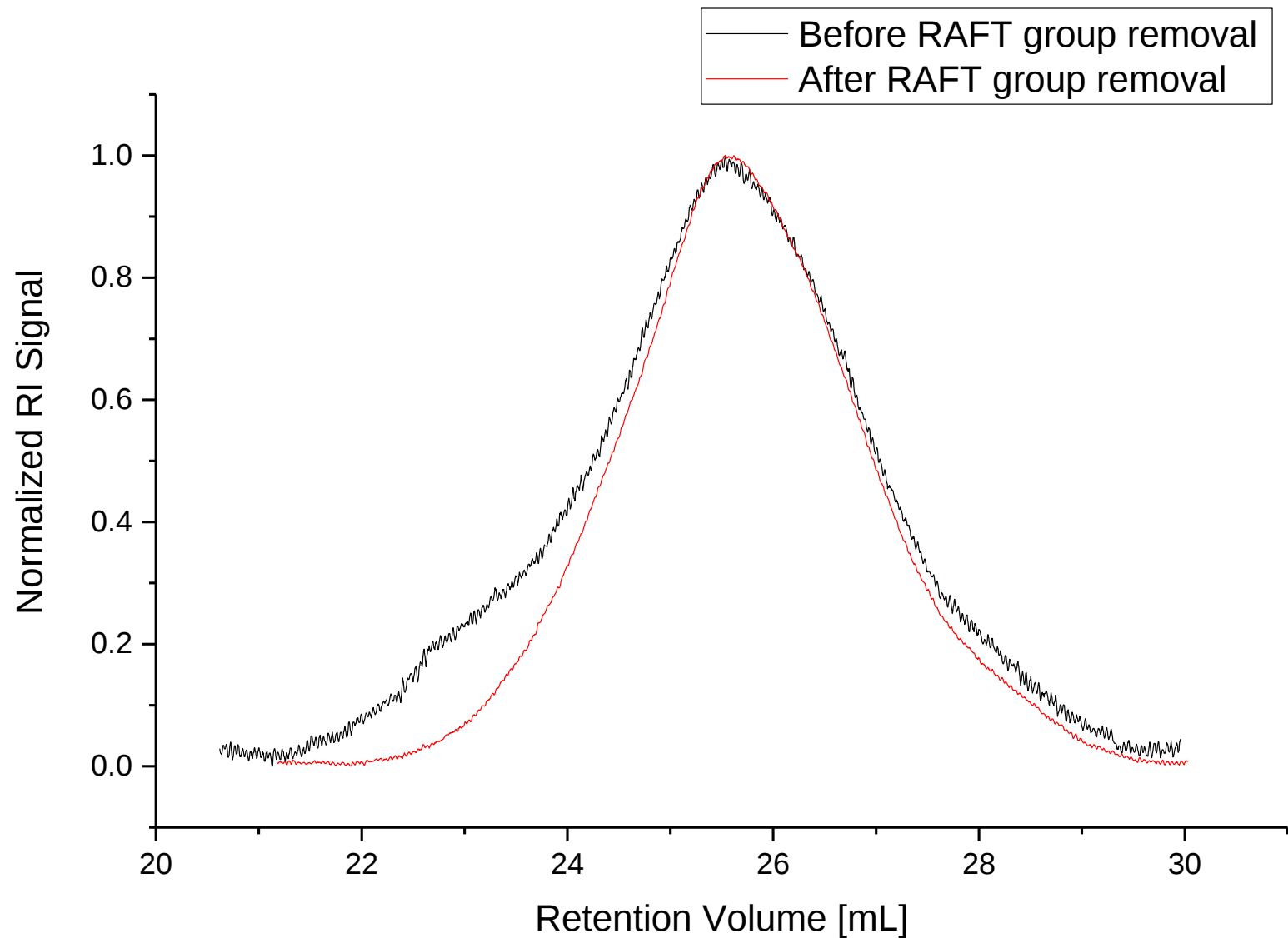


- NMR and SEC: no cross-linking or chemical changes in the polymer
- To do: verify/quantify via UV-Vis

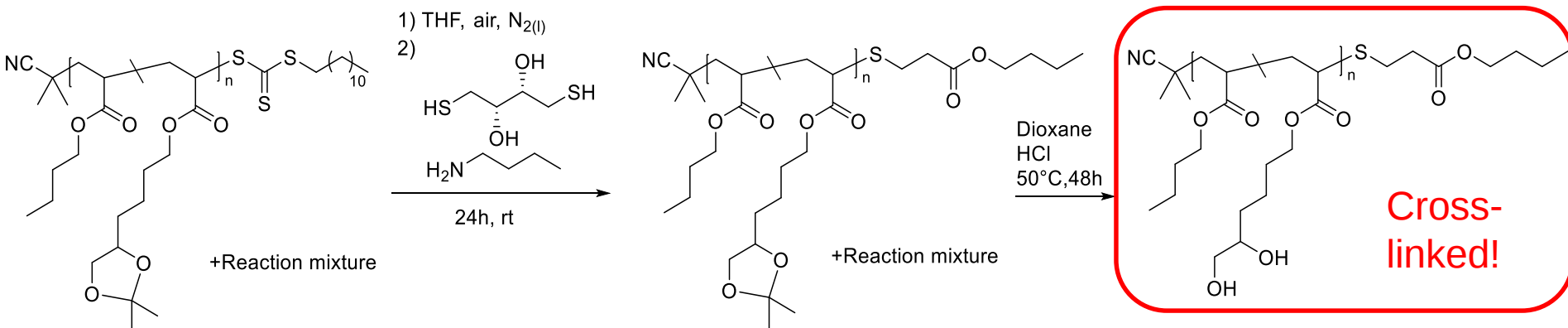
DORnet INITIAL TRAINING NETWORK



before RAFT removal



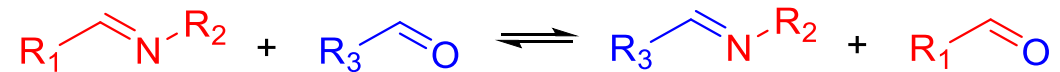
- Deprotection: One-Pot



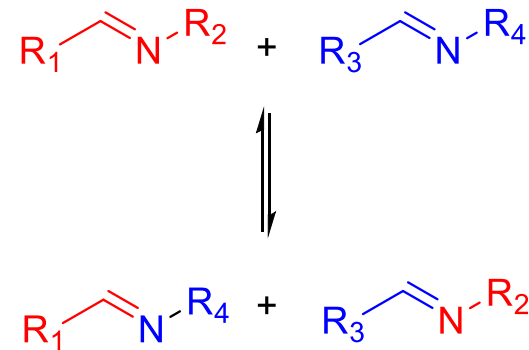
- Try to clean intermediate polymer first
- Try different acid, milder conditions

- Extendable network
- Employ dynamic bonds
- Imine exchange
- Imine: Aldehyde + Amine
- Works at elevated T
- 'Vitrimers Network'
- Maintains same number of crosslinks

Imine / aldehyde exchange:

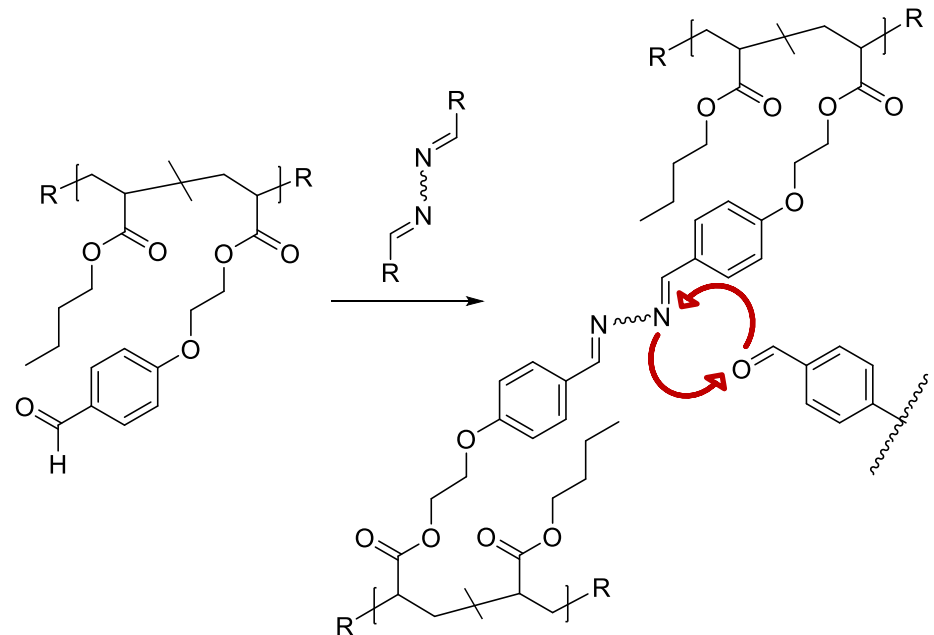
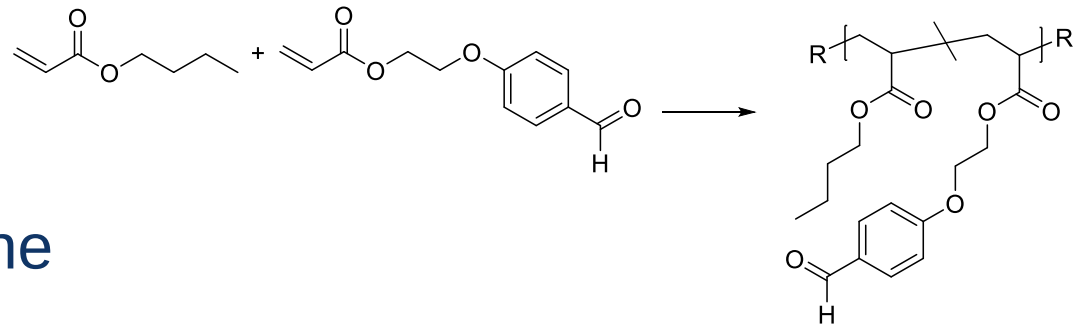


Imine exchange:

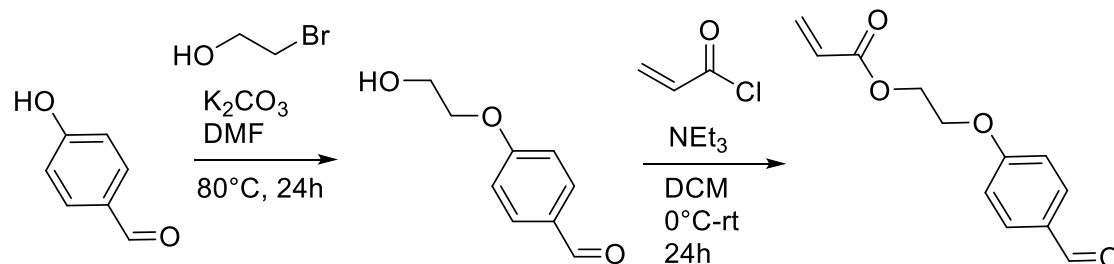


Exchange accelerates in presence of aldehydes

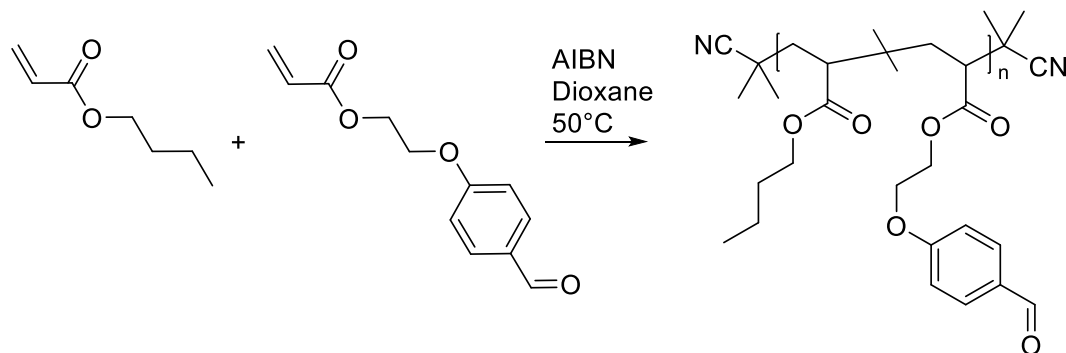
- Aldehyde functional polymer
- Cross-linking with diimine
- Low density of cross-links
 - DP 5000 (~500 000 Da)
 - ~10 links/chain (0.2% functionality)



- Two step synthesis

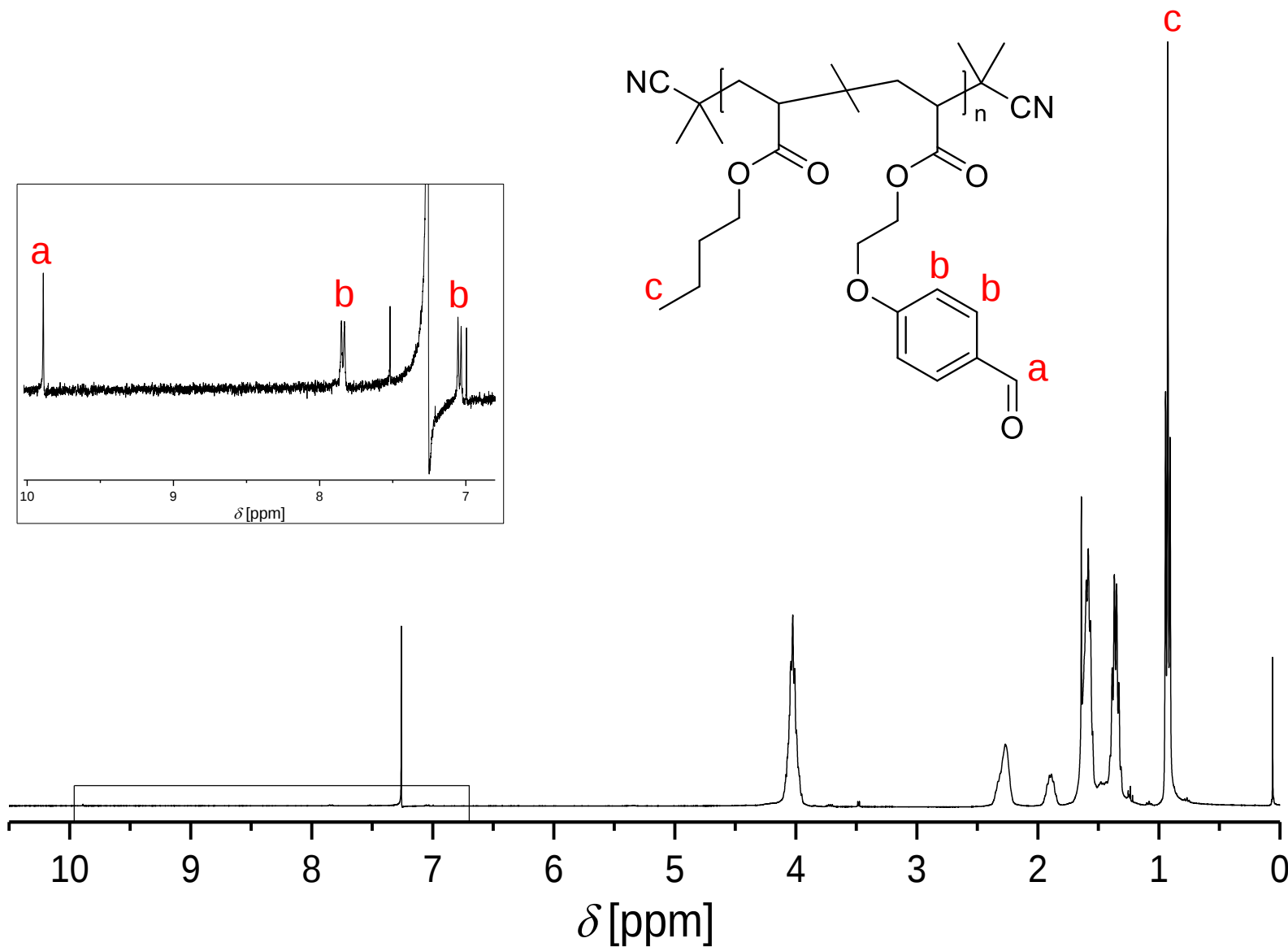


- Employment in free radical polymerization (FRP)



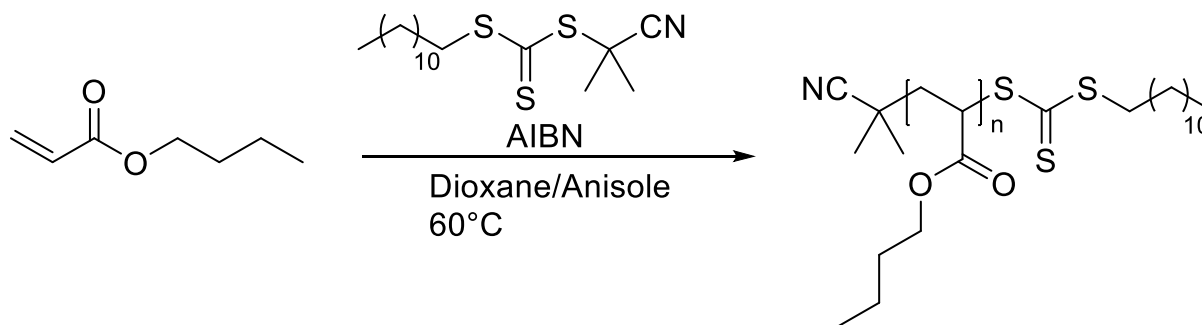
- $M_n = 300\,000\text{ Da}$; $\bar{D} = 2.16$, $\sim 0.2\%$ functionality

Aldehyde Copolymer NMR



- Experiments with FRP
- Best case: $M_n \sim 400\,000$ Da; $\bar{D} = 2.00$

- RAFT Polymerization



- Dioxane: $M_n \sim 200\,000$ Da; $\bar{D} \sim 2.00$
- Anisole: $M_n \sim 500\,000$ Da; $\bar{D} \sim 1.50$; but no clean SEC
- Work with FRP results
- ATRP? Color? Orthogonality to aldehyde?

- Make deprotection work
- Test cross-linking reactions
- Synthesize single and double networks
- Explore more systems (dynamic/ non-dynamic)
- Try to accomplish larger polymer sizes
- Mechanical testing