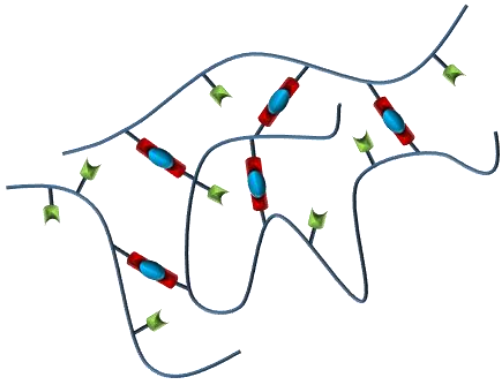
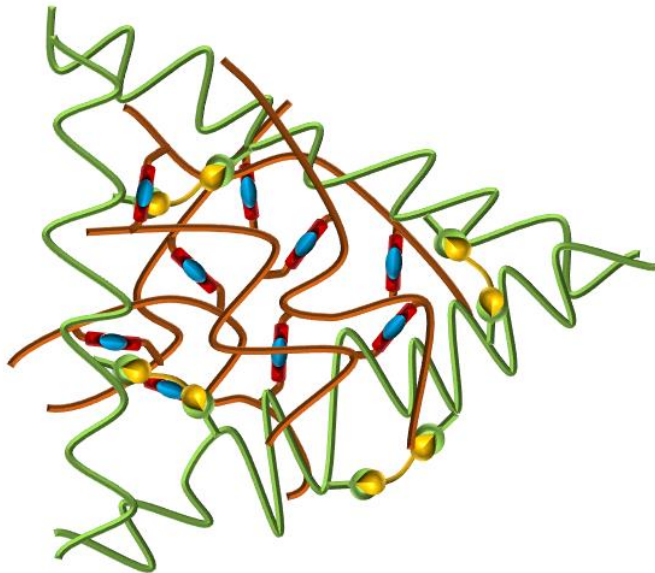




UPDATE COPENHAGEN

Larissa Hammer

Double Networks



Thermoplastic Elastomers

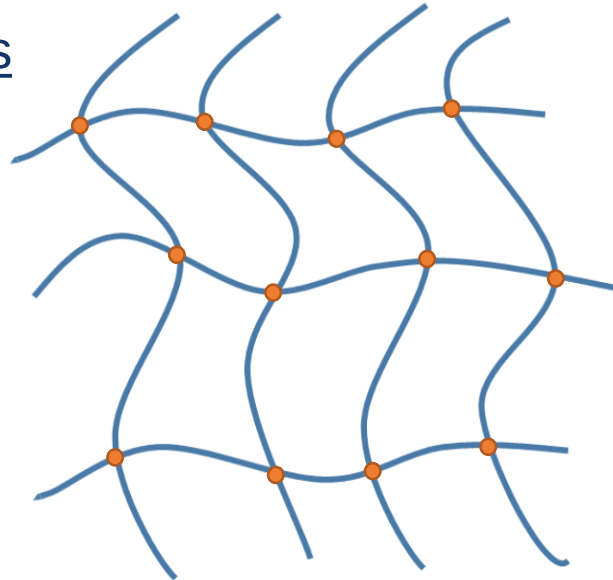


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- Network/gel: cross-linked polymer chains

Permanent cross-links

- creep resistance
- enhanced mechanical strength
- no coherence loss
- no processing
- no recycling



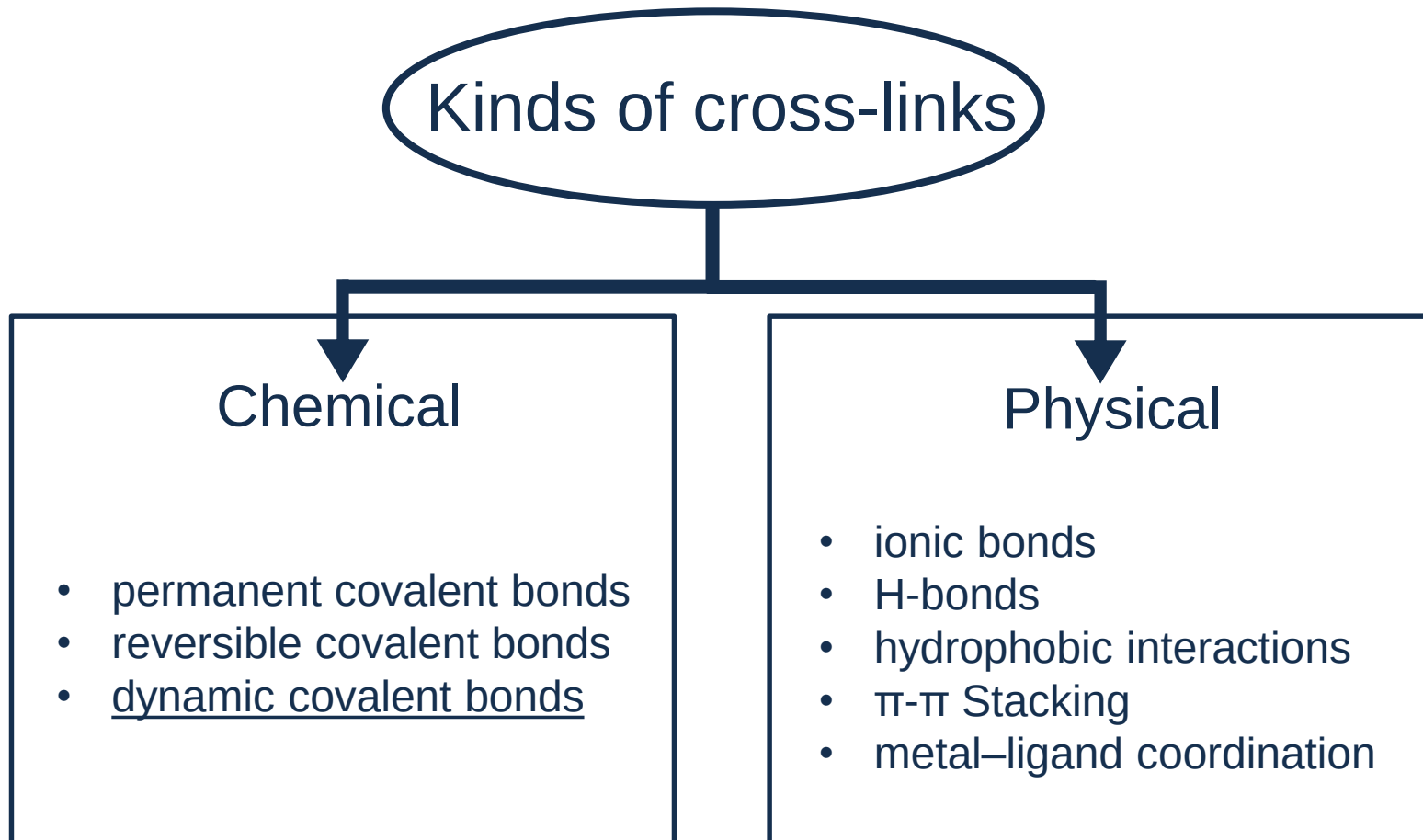
Non-permanent cross-links

- easily processable
- recyclable
- creep
- limited mechanical properties

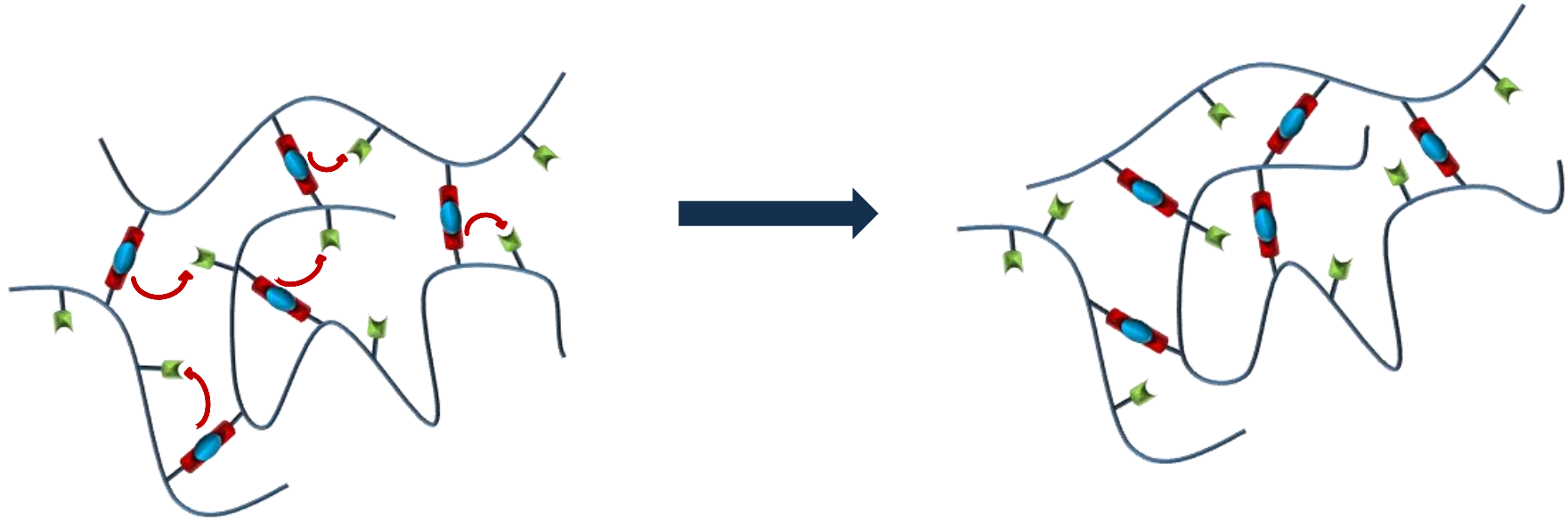


Combine the best of both

- Combination of at least two distinct dynamic modes
- 'Double' mechanical behaviour



- Employ dynamic 'vitrimeric' bonds in network



Vitrimeric bonds:

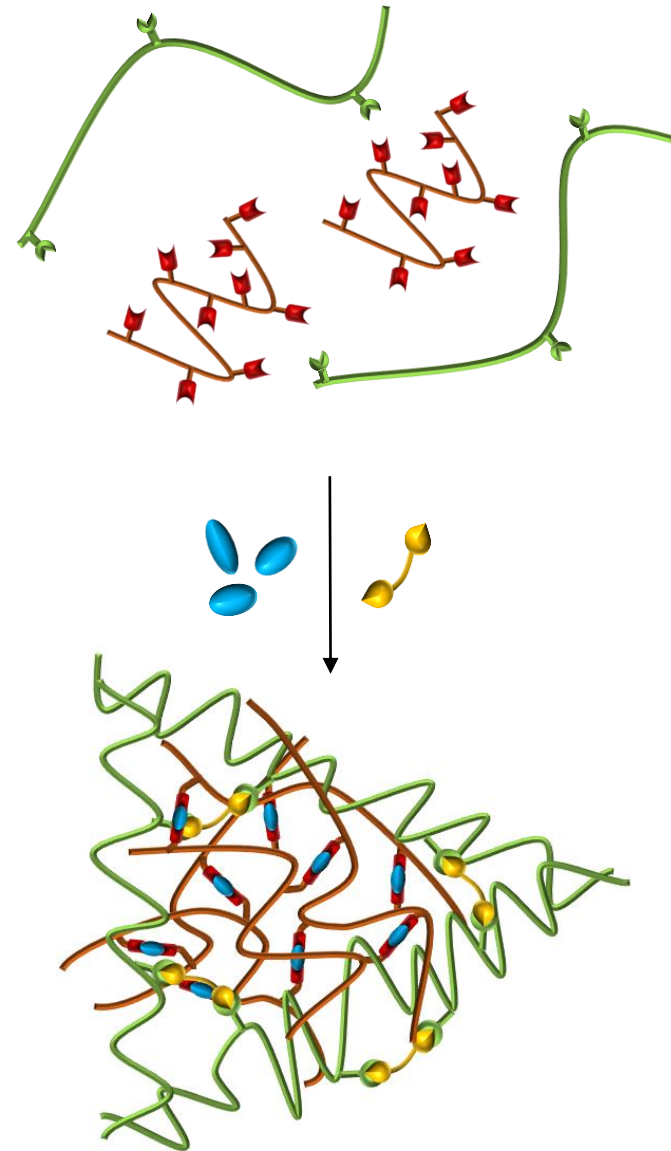
- Associative mechanism
- New linking point involved in reaction
- No bond breaking
- Same amount of cross-links



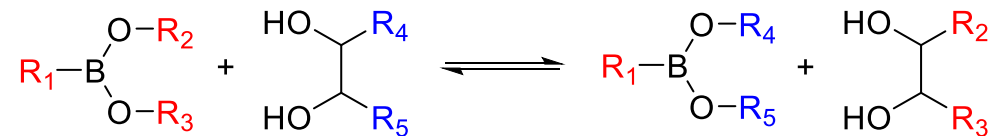
Material:

- Self-healing
- Malleable
- Reprocessable
- Maintains network integrity
- Tough

- Goal: create a elastomer Double Network based on dynamic covalent bonds
- Creating two interpenetrating network structures
 - Creating thermoplastics with functional groups
 - Orthogonal cross-linking
- Two independent modes

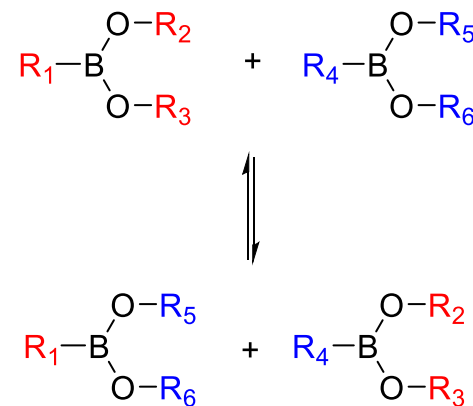


Boronic ester / diol exchange:



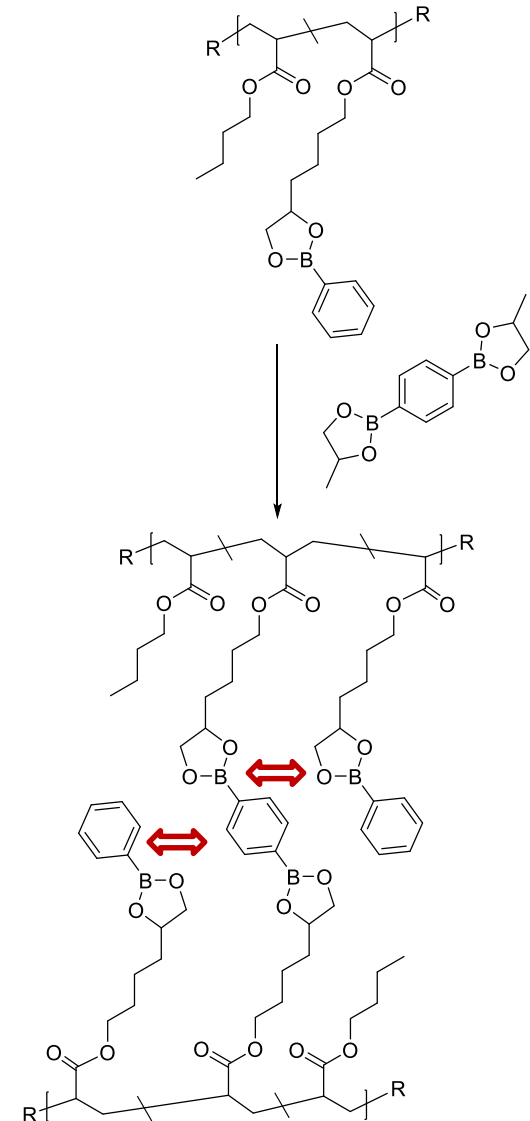
- ‘Vitrimers Network’
- Boronic ester exchange
- Works at rt/ in bulk

Boronic ester exchange:



Exchange
accelerates in
presence of
diols or
moisture

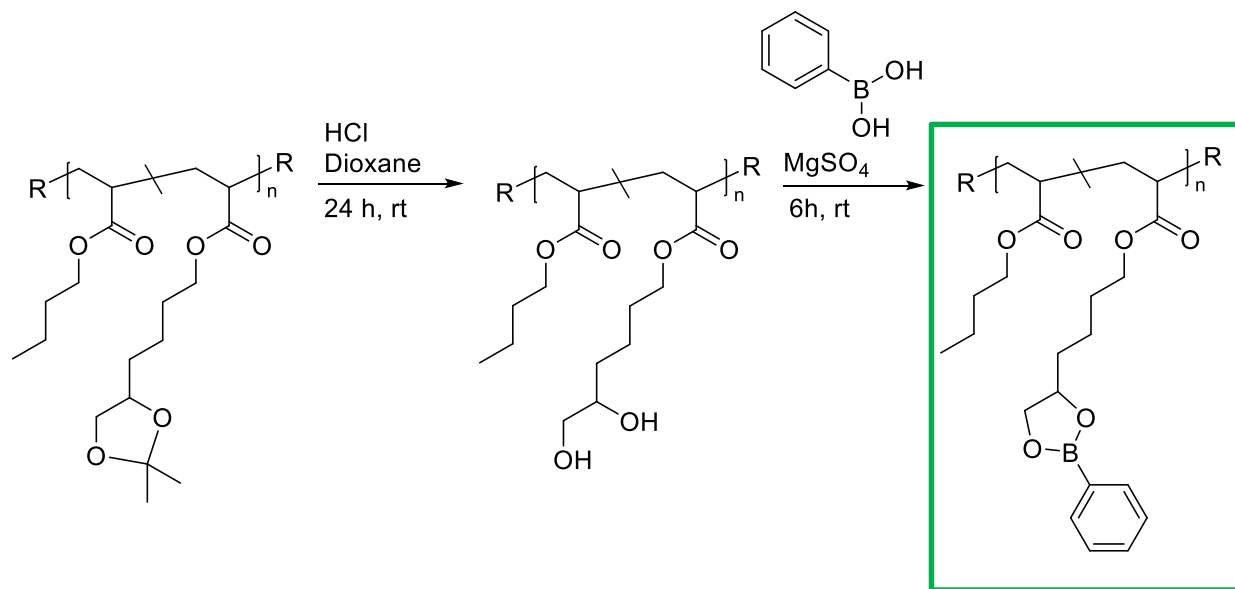
- Butyl acrylate system (low T_g)
- High density of cross links
 - DP 1000 (~100 000 Da) or lower
 - 10% functionality
- Diol-based polymer protected with boronic acid
- Cross-linking via di-boronic ester
- Rapid exchange
- Active at RT



-

- [illegible]

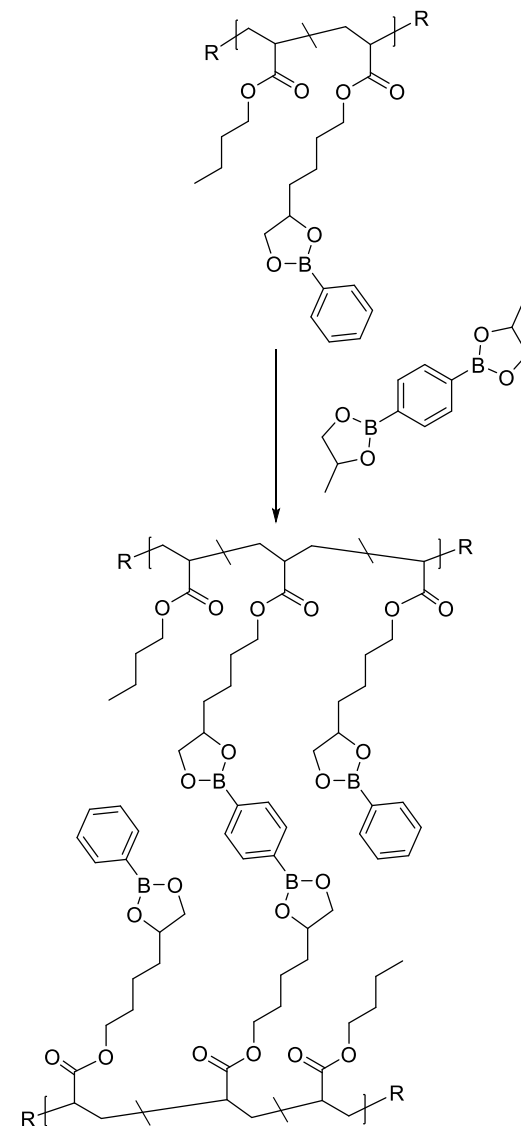
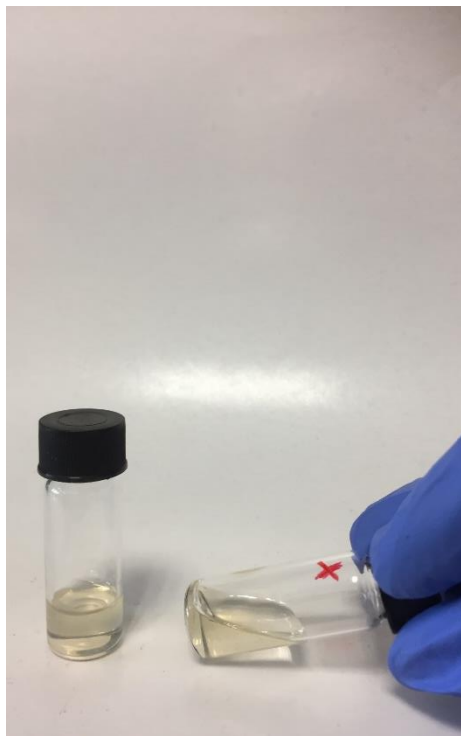
$M_n = 90$ kDa; $\bar{D} = 1.25$,
10% functionality,
>95% monomer conv.



- Thermoplastic precursor for Network 1

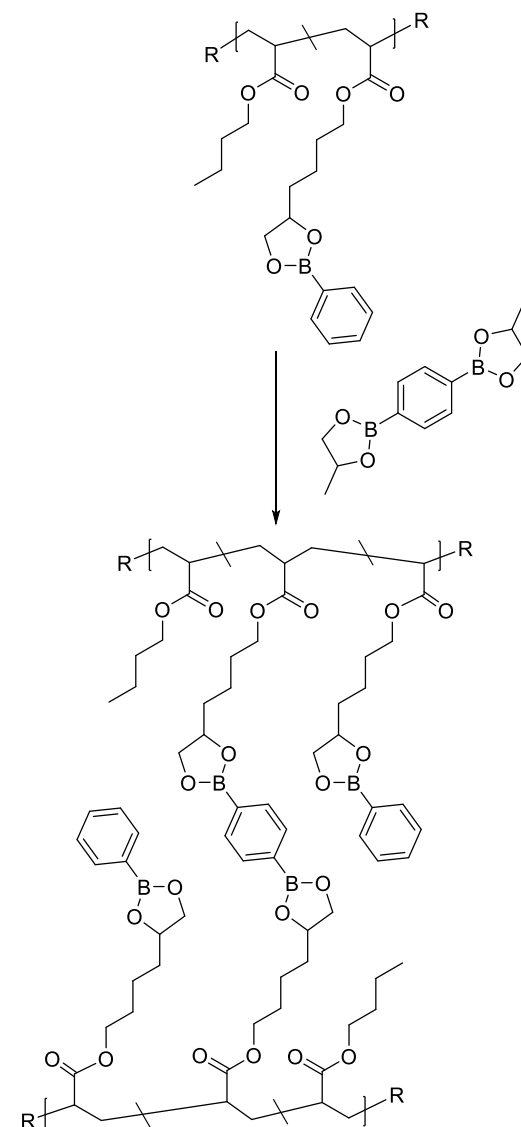
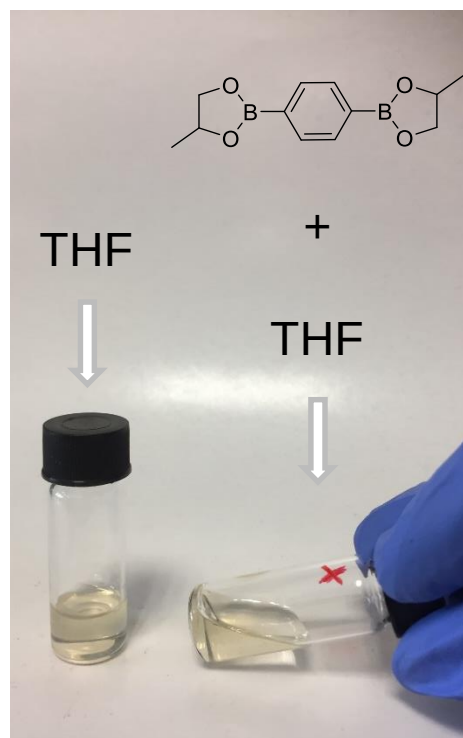
Cross-linking Test

- Solution of precursor in THF



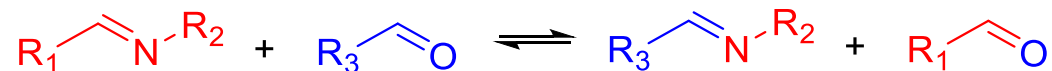
Cross-linking Test

- Solution of precursor in THF



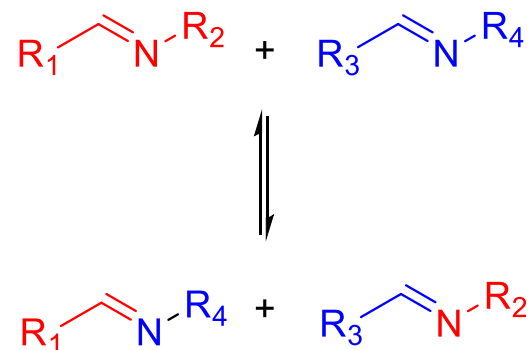
- Second dynamic mode

Imine / aldehyde exchange:



- 'Vitrimers Network'
- Imine exchange
- Imine = Amine + Aldehyde
- Slower than boronic ester exchange

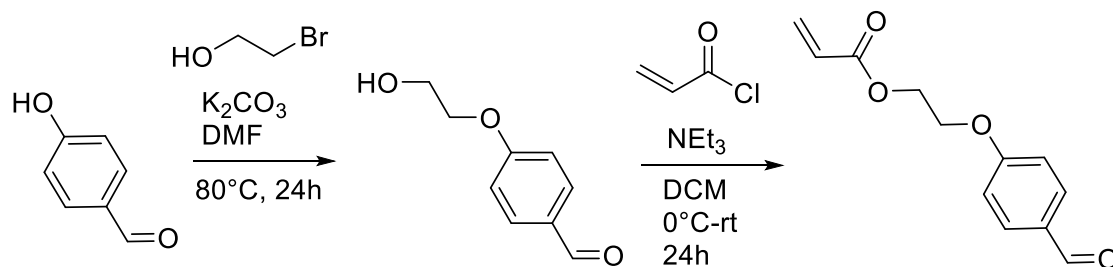
Imine exchange:



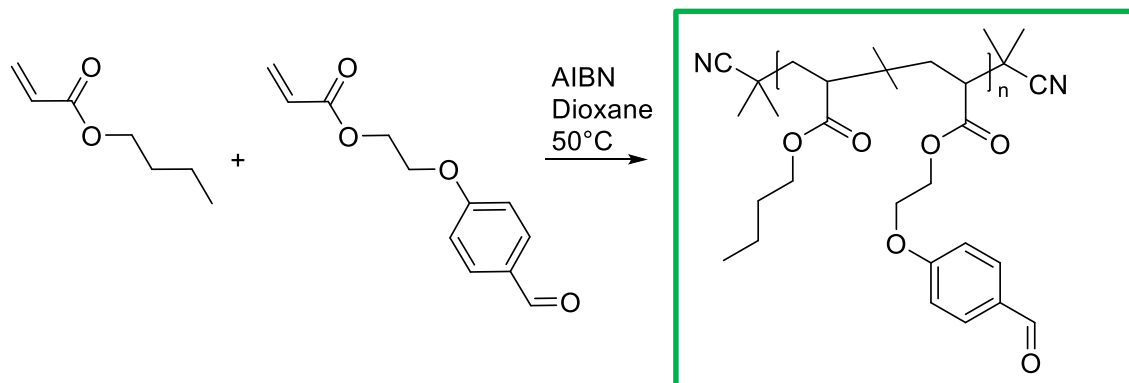
Exchange
accelerates in
presence of
aldehydes

-

- Two step synthesis



- Employment in free radical polymerization (FRP)

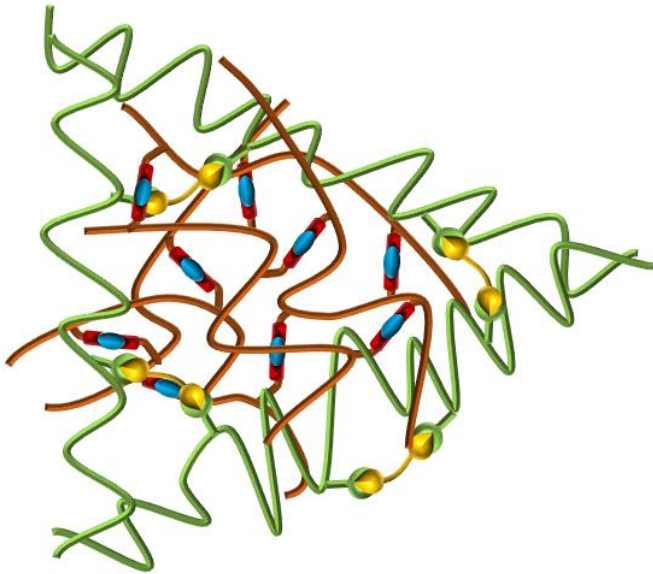


- $M_n = 300$ kDa; $\bar{D} = 2.16$, $\sim 0.2\%$ functionality
- Thermoplastic precursor for Network 2

- Verify cross-linking reactions by
 - IR spectroscopy
 - Swelling tests
 - Rheometry
- Combine both precursors
- Create Double Network
- Testing

- Use know-how to create new Double Networks
- Single Networks
- Connected Double Networks
- Integrate permanent bonds

Double Networks

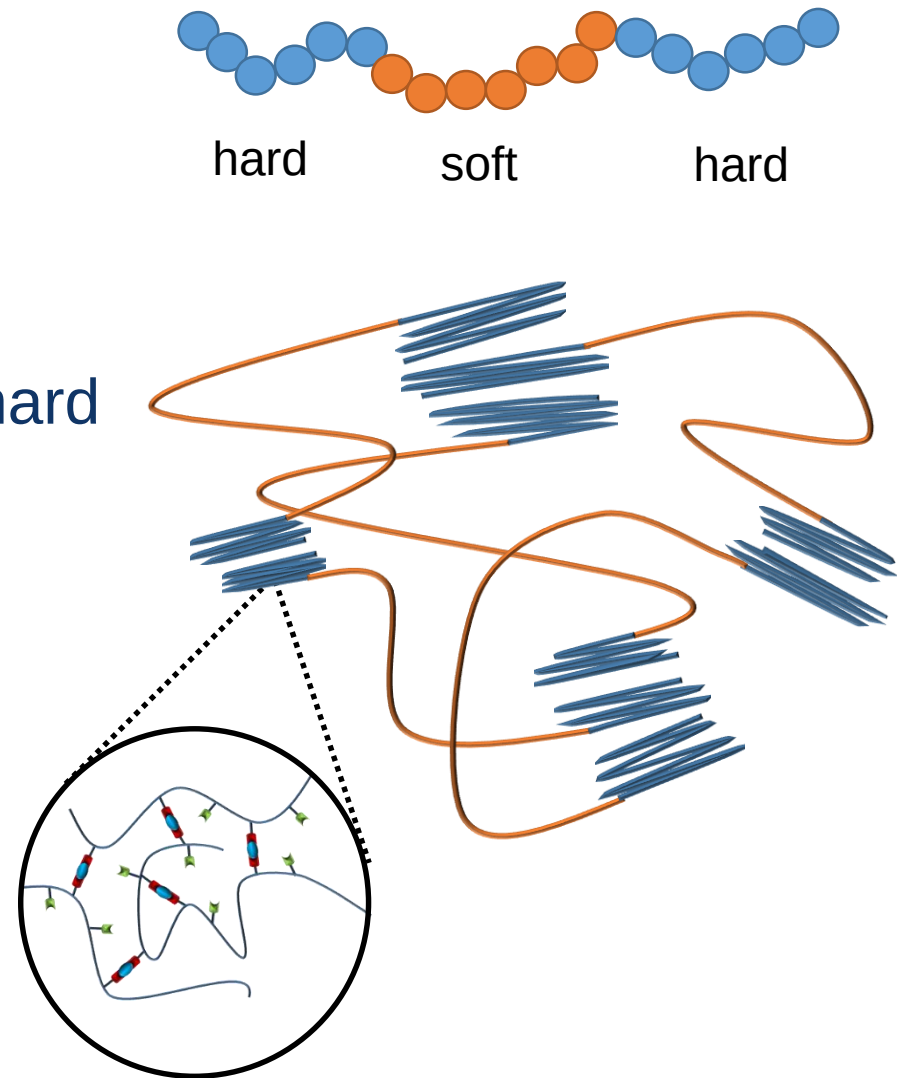


Thermoplastic Elastomers

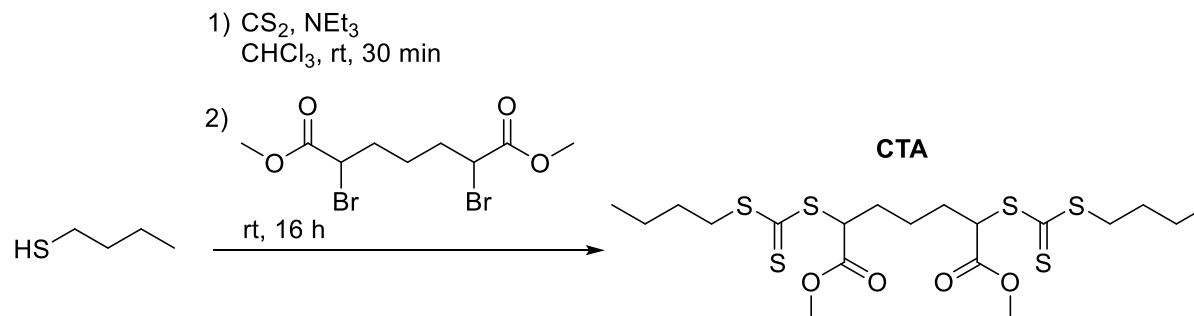


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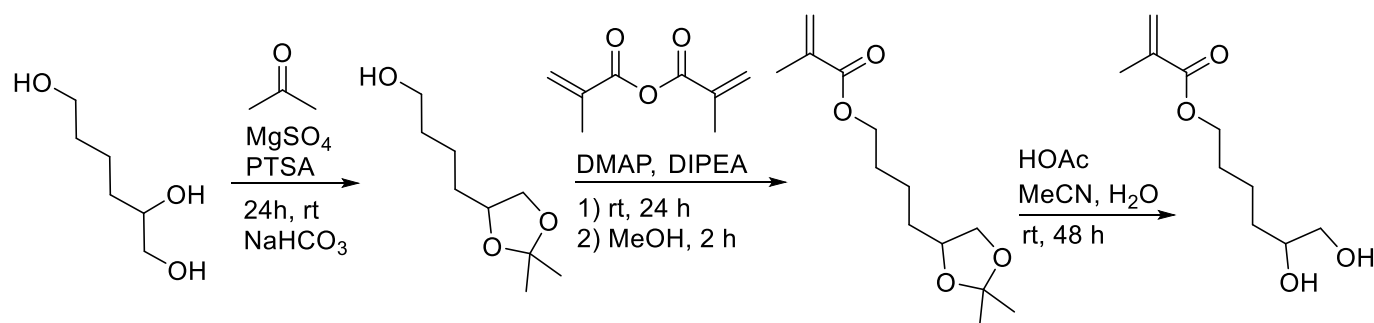
- Thermoplastic Elastomer
- A-B-A triblock co-polymer
- Hard-soft-hard phase
- Microphase separation
 - One dynamic below T_g of hard phase
- Decorate hard phase with dynamic covalent bonds
 - Second dynamic mode
 - Retain structure above T_g



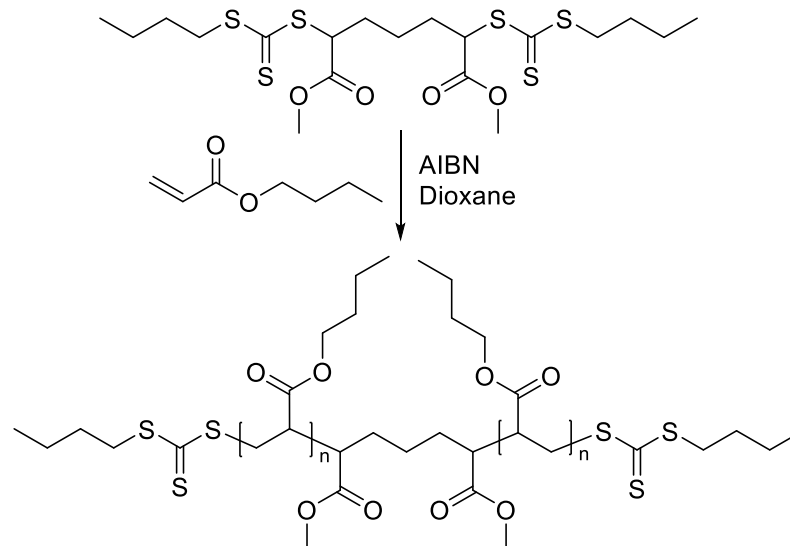
- Requires double functional symmetric CTA for RAFT polymerization



- Diol functional monomer for hard phase (MMA)

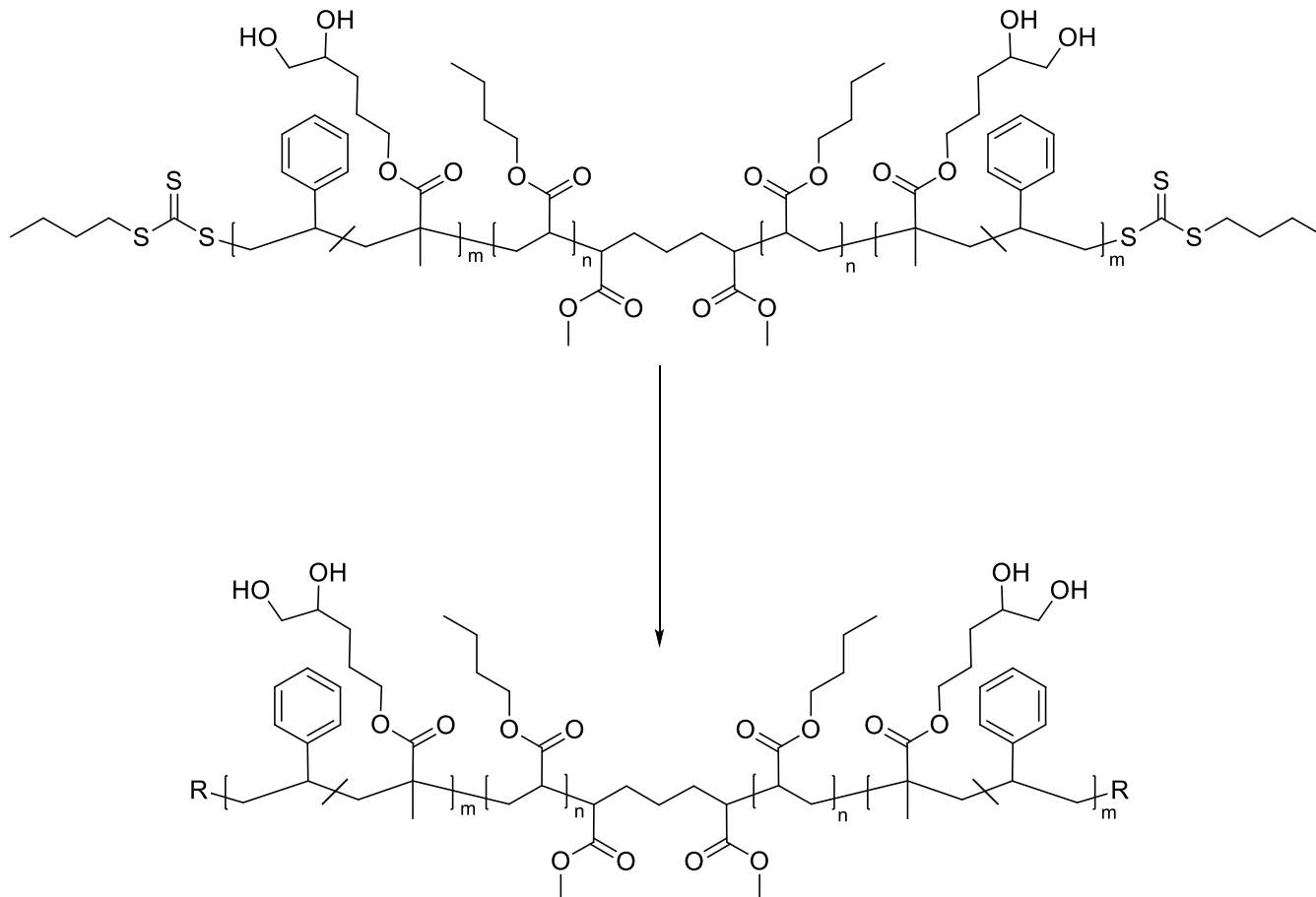


- Polymerization soft block
- Approx. 100 kDa, narrow PDI
- Kinetic studies

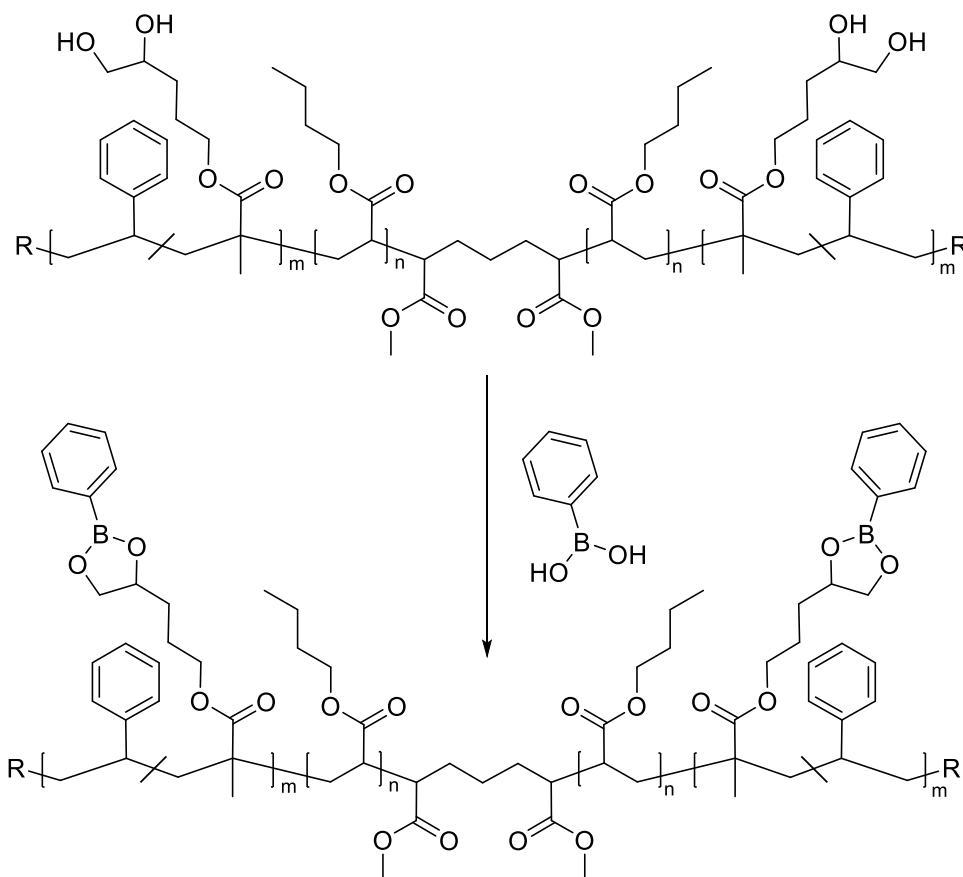


- ionality
-
- Chemical reaction scheme showing the synthesis of a telechelic poly(ester-ether) with terminal disulfide groups.
- The starting material is a poly(ester-ether) with terminal disulfide groups, reacting with styrene and a diol (1,6-hexanediol) in the presence of AIBN and Dioxane.
- The reaction conditions are AIBN and Dioxane.
- The product is a telechelic poly(ester-ether) with terminal disulfide groups, where the central chain is composed of repeating units derived from the starting material and the diol.

- RAFT group removal



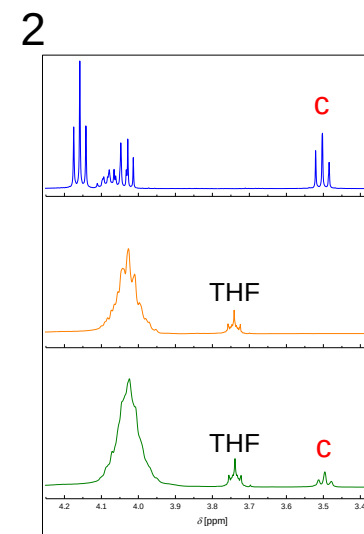
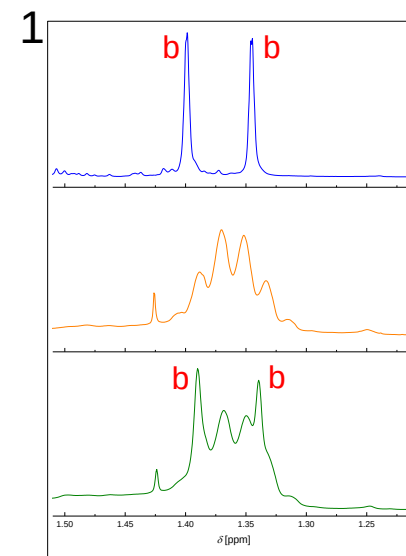
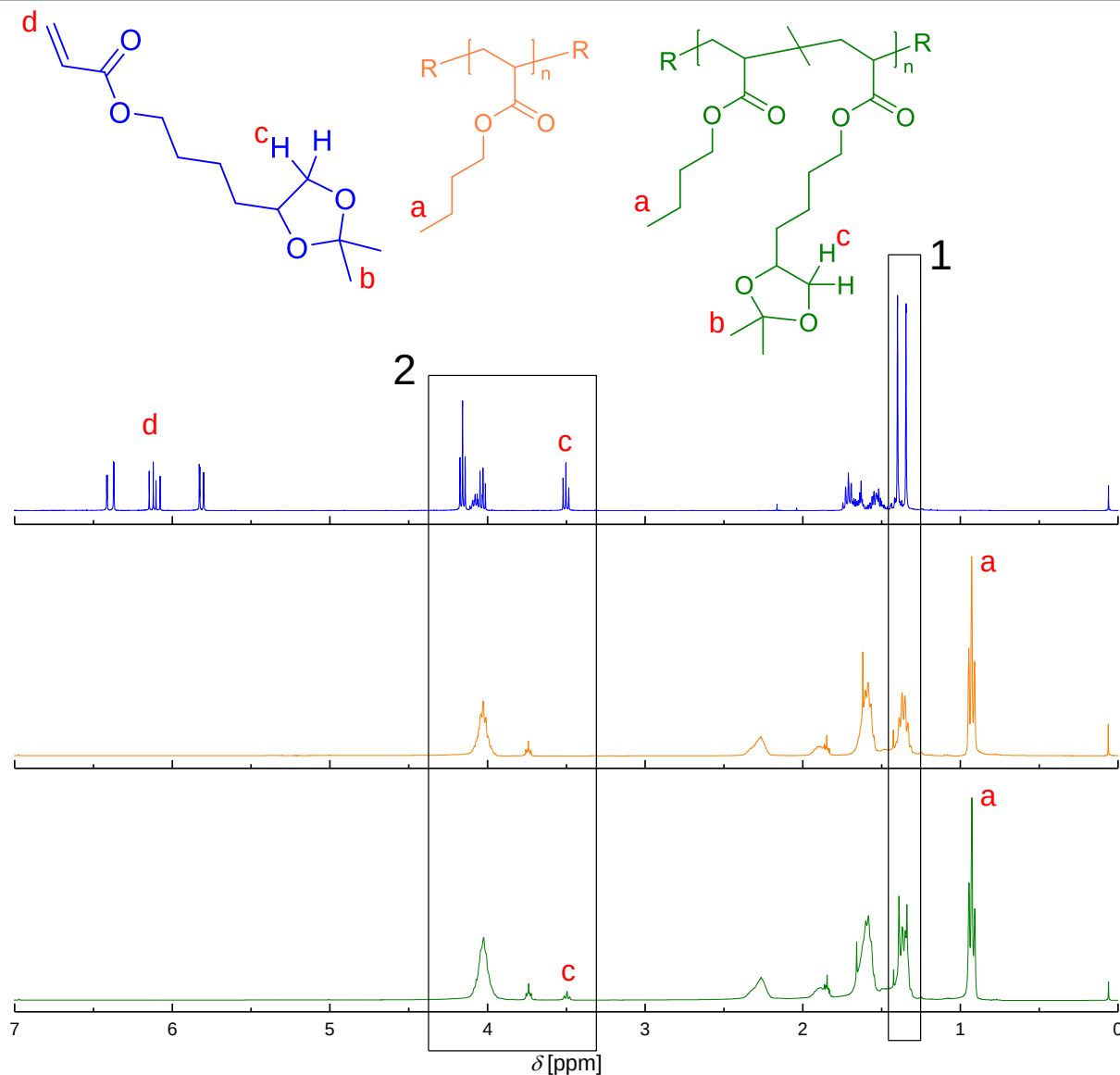
- Protection with phenylboronic acid

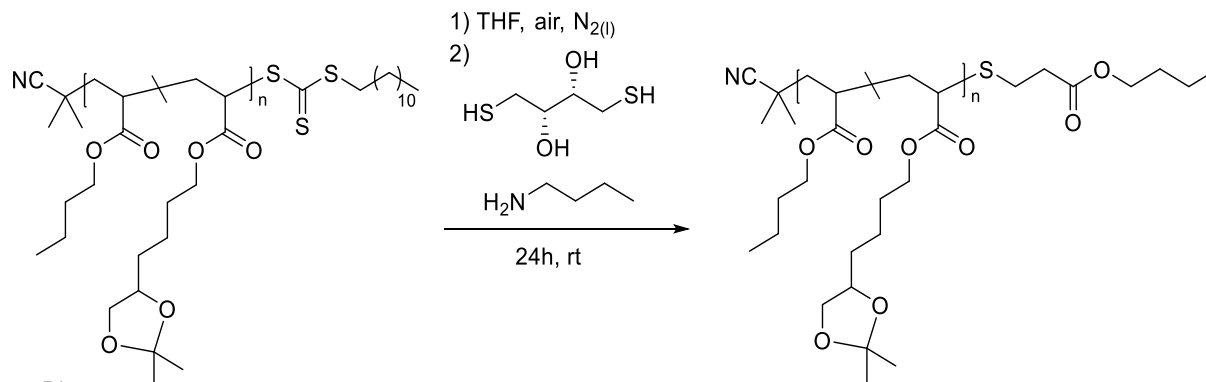


Thank you for your attention!



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+Dioxane
 +nBA

